

391. Topochemistry. Part IX.* The Crystal and Molecular Structures of *N*-5-Chlorosalicylideneaniline near 90 and 300°K.

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The crystal structure of *N*-5-chlorosalicylideneaniline has been solved by Patterson methods from partial three-dimensional, photographic data collected at 300 and 90°K. The co-ordinates of carbon, nitrogen, oxygen, and chlorine have been located by an anisotropic least-squares refinement with an accuracy of at least 0.006 and 0.007 Å at room and low temperature, respectively. The positional and isotropic temperature parameters of the hydrogen atoms attached to carbon have also been derived. The electron-density distribution in the O...N region has been determined by three-dimensional difference syntheses; at low temperature a single peak of 0.5 e/Å³ corresponds to hydrogen attached to oxygen; the room-temperature distribution shows two peaks (0.25 e/Å³ each), one of which is identified as the hydroxylic hydrogen.

The thermal vibrations have been analysed in terms of rigid-body motion. Molecular shape, packing arrangement, and bond lengths are discussed in the following Paper (Part X).

THE structure analyses of *N*-5-chlorosalicylideneaniline (I) † at room temperature and near the temperature of boiling nitrogen form part of a research programme on the photochemistry of the anils of salicylaldehyde. It has been shown (Part VI ¹) that the crystalline anils fall into two classes defined by their behaviour on irradiation with ultraviolet light (3600 Å). The first class develops an absorption band with a maximum near 4800 Å which can be eradicated thermally or by irradiation within this new band (reversible photochromy). To this class belongs 2-chloro-*N*-salicylideneaniline whose crystal structure will be described in the following Paper. The second class of solid anils does not colour in light, *i.e.*, is not photochromic; on the other hand, all members of this class show a more or less intense absorption band similar in position to the "photochromic" band mentioned above. In a sub-class characterised by a shortest axis of 4.6 ± 0.1 Å (see

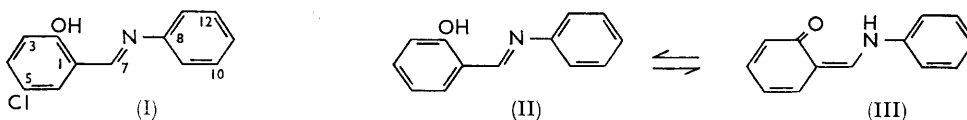


Table 2, Part VI), which includes *N*-5-chlorosalicylideneaniline, this band is sufficiently intense to impart to the crystals a red colour as against the yellow tinge of the majority of solid anils. This absorption band at 4800 Å has been shown to decrease with decreasing temperature and to vanish at the temperature of liquid air. The resultant colour change red \rightleftharpoons yellow is reversible in all compounds investigated (reversible thermochromy). Spectroscopic evidence presented in Parts VI and VIII is interpreted in terms of a tautomeric equilibrium (II) \rightleftharpoons (III).

It has further been shown that photochromy and thermochromy are mutually exclusive properties: they are determined not by the molecular species as such but by its crystal structure, for several anils occur in dimorphs of which one is thermochromic while the other is photochromic.

X-Ray structure analyses of salicylideneanilines have been carried out for two purposes: first, we compare the detailed features of the molecular structures and packing arrangements in thermochromic and photochromic crystals; secondly, we compare the structures of a thermochromic anil at two temperatures corresponding to bleaching and to maximum

* Part VIII, preceding Paper.

† Numbering shown is that used in the crystallographic descriptions and in bond dimensions.

¹ Part VI, *J.*, 1964, 2041.

coloration (low and high temperatures, respectively). Spectroscopic data indicate that in (I) the thermochromic band at 4800 Å has completely disappeared at 90°K while at room temperature conversion to the coloured form amounts to approximately 50% of the maximum obtainable just below the melting point (110°). The absolute concentration of the coloured species is, however, not known.

EXPERIMENTAL

N-5-Chlorosalicylideneaniline, C₁₃H₁₀ClON, *M* 231.7, crystallises from dioxan in red orthorhombic needles elongated along [010] and showing {100} and {001}. The cell dimensions were obtained at the two temperatures from high-order reflections on calibrated Weissenberg zero-level photographs; at room temperature: *a* = 12.319, *b* = 4.527, *c* = 19.48 Å; at low temperature: *a* = 12.175, *b* = 4.480, *c* = 19.25 Å (estimated error 0.1%). The changes in cell dimensions are no more than would be expected from normal thermal coefficients of expansion. Absences: 0*kl* for *l* odd, *h*0*l* for *h* odd. Space group: *Pca*2₁ or *Pcam*. *d*_{calc.}, for *n* = 4, 1.417 g./cm.³. *μ* = 30.0 cm.⁻¹; *F*(000) = 480. The higher-symmetry space group is ruled out by packing considerations; *Pca*2₁ was adopted, and proved correct by the eventual refinement.

Well-exposed photographs at either temperature show no signs of a superlattice nor any other abnormal features. Photographs of a large crystal taken with Mo-Kα radiation do not indicate any peculiarities in the range beyond the Cu-Kα sphere. Indeed, the similarities of photographs taken at the two temperatures suggest that any differences in structure produced by variations in temperature are minor.

Intensities of the *h*0*l*, *h*1*l*, *h*2*l*, and *h*3*l* levels were collected with Cu-Kα radiation on an equi-inclination Weissenberg camera with the multiple-film technique from a cylindrically-shaped crystal of diameter 0.29 mm.; the 0*kl* intensities were taken from a crystal cut to an approximate cylinder elongated along [100]. Low-temperature data were taken by means of the device described by Hirshfeld and Schmidt.²

The intensities (*hkl* and *hk̄l*) were estimated on two sets of six films exposed for approximately 24 and 3 hr., respectively; on the upper-level photographs only the extended spots were read. Reflections, "unobserved" on the strongest films were assigned intensities equal to the lowest value read on this film. Information on the intensity data is listed in Table 1. The averaged film readings were corrected for the Lorentz and polarisation factors, spot extension,³ and absorption.⁴

TABLE 1.
Intensity data.

	Low temperature		Room temperature	
	Reflections	Unobserved reflections	Reflections	Unobserved reflections
<i>h</i> 0 <i>l</i>	150	1	156	3
<i>h</i> 1 <i>l</i>	302	5	310	23
<i>h</i> 2 <i>l</i>	273	2	278	20
<i>h</i> 3 <i>l</i>	230	9	238	16
Total	955	17	982	62

Solution of the Structure.—The Patterson projection *P*(*xz*) was computed and yielded both the *x* parameter of the chlorine atom and approximate *x*, *z* parameters of the light atoms. The trial model was successfully refined by means of our standard least-squares programme (see Part V⁵). The weighting factors *w*(*hkl*) employed in this analysis varied approximately as *F*_o⁻³, but were modified in accordance with the number of observations of each reflection and their estimated reliabilities. "Unobserved" reflections were given weight ½ relative to observed reflections of equal magnitude; the programme included these reflections in the refinement cycle only if |*F*_o| > *kF*_{threshold}. The following scattering-factor curves were used: *f*_{Cl} Dawson,⁶ *f*_{C,N,O} Berghuis *et al.*,⁷ *f*_H McWeeny.⁸ No allowance was made for anomalous-dispersion effects of the chlorine atom.

² Hirshfeld and Schmidt, *Acta Cryst.*, 1956, **9**, 233.

³ Philips, *Acta Cryst.*, 1956, **9**, 819.

⁴ Bond, *International Tables for X-ray Crystallography*, Vol. II, ed. Kasper and Lonsdale, Kynoch Press, Birmingham, 1959, p. 295.

⁵ Part V, *J.*, 1964, 2030.

⁶ Dawson, *Acta Cryst.*, 1960, **13**, 403.

⁷ Berghuis, Haanappel, Potters, Loopstra, MacGillavry, and Veenendaal, *Acta Cryst.*, 1955, **8**, 478.

⁸ McWeeny, *Acta Cryst.*, 1951, **4**, 513.

During the first few cycles all atoms were treated isotropically; later, all hydrogens attached to carbon were located from a difference synthesis $\delta(xz)$ and inserted into the structure-factor calculations; subsequently, all atoms except hydrogen were assigned anisotropic temperature factors. The low-temperature $F(h0l)$ data were treated by an identical procedure based on the best x, z co-ordinates derived from the room-temperature data.

Next $P(yz)$ was computed to locate the y parameter of chlorine; however, overlap made this projection intractable. Attempts to determine y parameters from packing considerations alone also failed.

A generalised Patterson line synthesis along $(x, \frac{1}{2})$ (Fig. 1) was computed from the scaled $F^2(h1l)$, and interpreted as follows: this function contains the chlorine-chlorine vectors $(2x, 2y, \frac{1}{2})$ and $(\frac{1}{2}-2x, 0, \frac{1}{2})$ at $(2x, \frac{1}{2})$ and $(\frac{1}{2}-2x, \frac{1}{2})$ such that the height of the former peak (M) is modified by $\cos 2\pi 2y$ while the latter peak (U) is unmodified. A comparison of the two peak heights should therefore give y_{Cl} . However, while the two peaks were located on the line synthesis at the expected x co-ordinates they proved to lie in regions of high and low background,

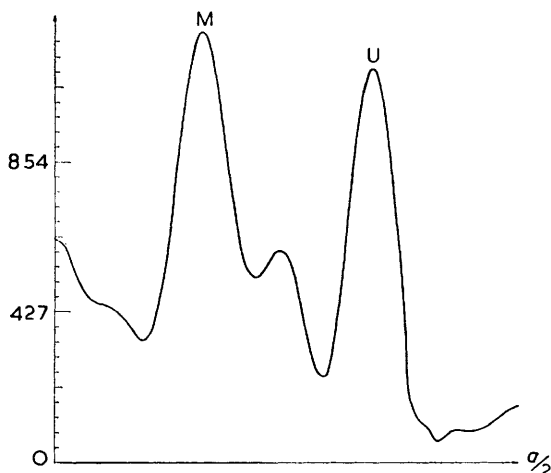


FIG. 1. *N*-5-Chlorosalicylideneaniline; generalised Patterson line synthesis $P_{k=1}(x, \frac{1}{2})$ in absolute units.

respectively. Rather than apply graphical corrections we chose the following analytical procedure: it was noted that M was surrounded by unmodified vectors due to c -glide plane related carbon atoms (1,2,3,5,7,8) while U was overlapped by unmodified vectors due to carbons 1 and 9, as well as by modified vectors due to the screw-axis related carbon atoms 1,2,3,5,7,8. (Vectors due to carbons 4,6,10,12,13, oxygen, and nitrogen fell outside the regions of U and M and could be ignored.) The carbon-vector contributions to the height of M, being independent of their y parameters, could be calculated from theoretical temperature-corrected carbon-carbon vector profiles (with temperature factors derived from the two-dimensional refinement); these were subtracted from the observed peak M to yield the "pure" modified peak M'. Since U contained y_C -dependent contributions which could not be allowed for at this stage, it was replaced by the theoretical profile U' corrected by the chlorine temperature factor. Comparison of M' and U' yielded $y_{Cl} = \pm 0.034$ (to be compared with the value of 0.0315 after refinement).

The value of -0.034 was chosen arbitrarily. Next, the $(h0l)$ projection was fitted to a molecular model which suggested the essentially coplanar arrangement of the two benzene rings. The heights of the light atoms, relative to chlorine at -0.034 were deduced from this model, and two sets of y co-ordinates calculated for the molecular orientations relative to the positive b direction. Rather than discriminate between the two sets of co-ordinates by structure-factor calculations we returned once more to the Patterson line synthesis. Since the (alternative) contributions of the modified carbon-carbon vectors, corresponding to the two sets of y_C -co-ordinates, could now be computed the observed peak height U, composed of U', the y_C -independent carbon-carbon vectors, and either of the two modified carbon vector sets, provided a choice for the molecular orientation. The numerical computations indicated an acute molecular tilt

to the positive b direction; the corresponding co-ordinates were inserted into structure-factor calculations which confirmed the correctness of this choice.

Positions of the hydrogens attached to carbon were computed on the basis of C-H bond lengths of 1.08 Å and of normal bond angles. A series of thermally isotropic cycles, in which the hydrogen parameters were not allowed to move, was followed by anisotropic cycles in which the positional and isotropic thermal parameters of the hydrogens were also adjusted. Refinement was stopped when all residual shifts had been reduced to values equal or less than one tenth of the corresponding standard deviations. The refinement of the room-temperature data followed a strictly analogous procedure. The "final" agreement factors are: $r = \Sigma w(k^2 F_o^2 - |F_c|^2) / \Sigma w k^4 F_o^4 = 0.014$; $R = \Sigma(|kF_o - |F_c||) / \Sigma k F_o = 0.057$ for all reflections excepting $|F_c| < |kF_{\text{threshold}}|$; at low temperature, $r = 0.023$; $R = 0.062$.

RESULTS

An electron-density projection $\rho(xz)$ of the room-temperature structure is shown in Fig. 2; two reflections, $F(200)$ and $F(203)$, believed to suffer from secondary extinction were included in the summation as F_c rather than F_o .

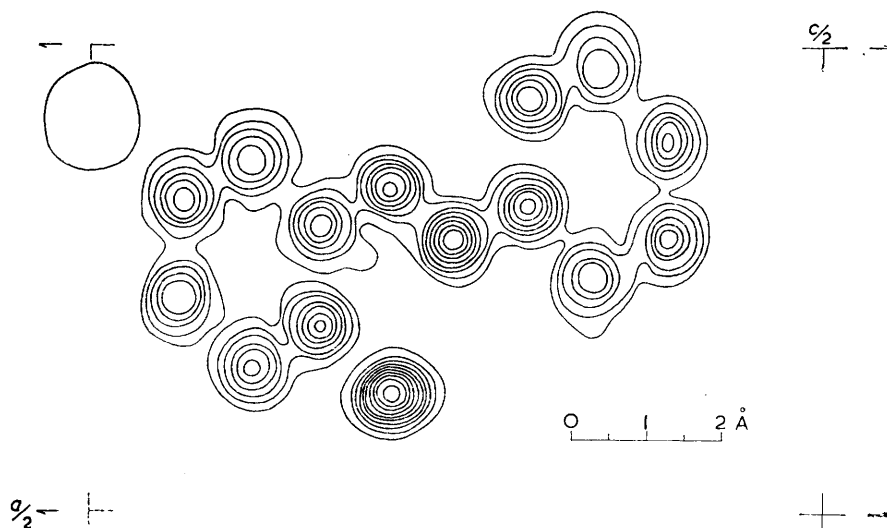


FIG. 2. *N*-5-Chlorosalicylideneaniline; electron-density projection $\rho(xz)$. Contour interval 1 e/Å²; lowest contour 2 e/Å². Contours on the chlorine atom are omitted.

TABLE 2.
Atomic co-ordinates (Å).

Atom	x	y	z	x	y	z	Atom	x	y	z	x	y	z
	Low temperature			Room temperature				Low temperature			Room temperature		
C ₁	2.352	2.061	3.081	2.371	2.106	3.069	N ...	2.575	3.622	4.859	2.584	3.649	4.858
C ₂	3.715	1.674	3.082	3.743	1.749	3.068	O ...	4.577	2.224	3.984	4.595	2.296	3.961
C ₃	4.206	0.761	2.155	4.219	0.821	2.137	Cl ...	0.956	-0.141	0.000	0.966	-0.085	0.000
C ₄	3.365	0.205	1.214	3.387	0.288	1.198	H ₃ ...	5.176	0.504	2.209	5.232	0.648	2.177
C ₅	2.011	0.564	1.207	2.025	0.632	1.202	H ₁ ...	3.651	-0.524	0.479	3.708	-0.496	0.575
C ₆	1.500	1.468	2.123	1.526	1.504	2.118	H ₆ ...	0.578	1.766	2.093	0.601	1.790	2.120
C ₇	1.811	3.007	4.020	1.838	3.050	4.022	H ₇ ...	0.859	3.273	3.937	0.930	3.232	3.971
C ₈	2.065	0.087	5.778	2.078	0.069	5.785	H ₉ ...	-0.047	0.060	5.211	0.024	-0.013	5.195
C ₉	0.719	0.523	5.821	0.737	0.499	5.829	H ₁₀ ...	-0.608	1.744	6.846	-0.544	1.668	6.877
C ₁₀ ...	0.335	1.444	6.760	0.348	1.399	6.782	H ₁₁ ...	0.913	2.681	8.446	0.905	2.516	8.408
C ₁₁ ...	1.241	1.996	7.675	1.259	1.935	7.679	H ₁₂ ...	3.221	1.971	8.329	3.160	1.787	8.282
C ₁₂ ...	2.584	1.573	7.623	2.572	1.543	7.626	H ₁₃ ...	3.968	0.530	6.616	4.026	0.589	6.575
C ₁₃ ...	2.982	0.642	6.678	3.000	0.608	6.670							

To locate the hydrogen atom (not included at any stage in the least-squares calculations) in the O...N region we computed difference syntheses $\delta(xz)$ from $\Delta F(h0l)$ with phase angles of the three-dimensional refinement; these were followed by three-dimensional plane sections with $\Delta F(hkl)$ (Figs. 3 and 4) parallel to [010] whose x, z parameters corresponded to the maxima in $\delta(xz)$ in the O...N region. The peak in the low-temperature projection $\delta(xz)$ is fairly round with a height of $0.5 \text{ e}/\text{\AA}^2$; the corresponding three-dimensional plane sections (Fig. 3) show a maximum of approximately $0.5 \text{ e}/\text{\AA}^3$; the centre of this peak lies at 1.03 \AA from the oxygen, such that the line joining this peak to the oxygen makes an angle of 104° with the O-C₂ bond; the peak lies 0.1 \AA above the best plane defined by

TABLE 3.
Standard deviations (\AA) of atomic co-ordinates.

Atom	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$	Atom	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$
Low temperature						Room temperature							
C ₁	0.0047	0.0053	0.0044	0.0036	0.0046	0.0040	N	0.0036	0.0053	0.0033	0.0028	0.0036	0.0033
C ₂	0.0042	0.0065	0.0041	0.0033	0.0049	0.0041	O	0.0029	0.0039	0.0034	0.0029	0.0036	0.0037
C ₃	0.0039	0.0067	0.0043	0.0038	0.0051	0.0045	Cl	0.0015	0.0013		0.0010	0.0014	
C ₄	0.0045	0.0059	0.0044	0.0041	0.0050	0.0045	H ₃	0.046	0.083	0.062	0.040	0.050	0.051
C ₅	0.0038	0.0072	0.0045	0.0038	0.0052	0.0041	H ₄	0.065	0.094	0.064	0.041	0.054	0.050
C ₆	0.0037	0.0064	0.0043	0.0033	0.0046	0.0039	H ₅	0.057	0.087	0.062	0.033	0.041	0.037
C ₇	0.0040	0.0065	0.0045	0.0037	0.0047	0.0040	H ₇	0.071	0.087	0.079	0.054	0.060	0.063
C ₈	0.0041	0.0058	0.0060	0.0039	0.0041	0.0042	H ₉	0.046	0.059	0.050	0.074	0.055	0.057
C ₉	0.0041	0.0069	0.0044	0.0040	0.0052	0.0047	H ₁₀	0.082	0.102	0.084	0.056	0.061	0.067
C ₁₀	0.0044	0.0062	0.0043	0.0046	0.0052	0.0051	H ₁₁	0.066	0.097	0.085	0.068	0.060	0.085
C ₁₁	0.0055	0.0064	0.0044	0.0054	0.0056	0.0046	H ₁₂	0.050	0.053	0.053	0.047	0.046	0.052
C ₁₂	0.0045	0.0068	0.0043	0.0050	0.0057	0.0047	H ₁₃	0.073	0.111	0.089	0.055	0.069	0.065
C ₁₃	0.0043	0.0067	0.0043	0.0047	0.0050	0.0046							

TABLE 4.
Thermal parameters (\AA^2).

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{23}	U_{13}	Atom	U
Low temperature								
C ₁	0.0095	0.0068	0.0116	-0.0038	0.0054	0.0009	H ₃	-0.014
C ₂	0.0122	0.0085	0.0109	0.0002	0.0009	-0.0006	H ₄	0.004
C ₃	0.0143	0.0122	0.0103	0.0031	0.0024	0.0044	H ₆	-0.009
C ₄	0.0178	0.0039	0.0114	-0.0017	0.0029	0.0017	H ₇	0.015
C ₅	0.0099	0.0087	0.0102	-0.0006	-0.0034	-0.0006	H ₉	-0.026
C ₆	0.0114	0.0071	0.0149	0.0005	0.0017	-0.0015	H ₁₀	0.015
C ₇	0.0116	0.0062	0.0145	0.0003	-0.0002	-0.0001	H ₁₁	0.008
C ₈	0.0102	0.0090	0.0195	0.0003	-0.0009	-0.0005	H ₁₂	-0.021
C ₉	0.0099	0.0116	0.0133	0.0018	-0.0015	-0.0012	H ₁₃	0.018
C ₁₀	0.0150	0.0066	0.0149	0.0063	0.0015	0.0025		
C ₁₁	0.0204	0.0154	0.0107	0.0009	0.0071	0.0056		
C ₁₂	0.0179	0.0062	0.0118	-0.0004	0.0016	-0.0018		
C ₁₃	0.0157	0.0067	0.0133	-0.0006	0.0054	-0.0017		
N	0.0100	0.0152	0.0082	0.0014	0.0035	0.0011		
O	0.0068	0.0129	0.0149	-0.0038	-0.0038	-0.0018		
Cl	0.0109	0.0145	0.0089	-0.0021	-0.0032	-0.0013		
Room temperature								
C ₁	0.0328	0.0396	0.0362	0.0013	0.0062	0.0039	H ₃	0.012
C ₂	0.0373	0.0402	0.0384	-0.0032	0.0028	-0.0002	H ₄	0.018
C ₃	0.0379	0.0533	0.0484	0.0098	-0.0020	0.0112	H ₆	-0.007
C ₄	0.0421	0.0535	0.0418	0.0036	-0.0030	0.0058	H ₇	0.035
C ₅	0.0381	0.0439	0.0366	-0.0063	-0.0016	-0.0032	H ₉	0.032
C ₆	0.0313	0.0410	0.0384	-0.0028	0.0007	-0.0023	H ₁₀	0.038
C ₇	0.0318	0.0361	0.0460	-0.0026	0.0072	0.0023	H ₁₁	0.056
C ₈	0.0402	0.0417	0.0391	-0.0011	0.0050	0.0053	H ₁₂	0.016
C ₉	0.0373	0.0388	0.0518	0.0039	0.0005	0.0032	H ₁₃	0.042
C ₁₀	0.0482	0.0475	0.0561	0.0096	0.0058	0.0120		
C ₁₁	0.0688	0.0406	0.0432	0.0058	0.0012	0.0144		
C ₁₂	0.0619	0.0504	0.0433	-0.0038	-0.0014	-0.0041		
C ₁₃	0.0481	0.0404	0.0481	-0.0029	0.0044	-0.0035		
N	0.0297	0.0357	0.0386	0.0015	-0.0004	0.0011		
O	0.0279	0.0760	0.0572	0.0010	-0.0119	-0.0051		
Cl	0.0510	0.0629	0.0414	-0.0105	-0.0095	-0.0042		

TABLE 5.

Standard deviations (in Å²) of thermal parameters.

Atom	$\sigma(U_{11})$	$\sigma(U_{22})$	$\sigma(U_{33})$	$\sigma(U_{12})$	$\sigma(U_{23})$	$\sigma(U_{13})$	Atom	$\sigma(U)$
Low temperature								
C ₁	0.0016	0.0035	0.0017	0.0017	0.0019	0.0014	H ₃	0.012
C ₂	0.0016	0.0039	0.0016	0.0020	0.0022	0.0014	H ₄	0.016
C ₃	0.0016	0.0037	0.0017	0.0018	0.0022	0.0015	H ₆	0.012
C ₄	0.0019	0.0038	0.0016	0.0019	0.0019	0.0016	H ₇	0.024
C ₅	0.0015	0.0035	0.0016	0.0018	0.0021	0.0011	H ₉	0.012
C ₆	0.0016	0.0037	0.0016	0.0017	0.0021	0.0015	H ₁₀	0.020
C ₇	0.0015	0.0037	0.0017	0.0019	0.0022	0.0015	H ₁₁	0.020
C ₈	0.0019	0.0043	0.0024	0.0018	0.0021	0.0013	H ₁₂	0.012
C ₉	0.0014	0.0036	0.0017	0.0020	0.0022	0.0014	H ₁₃	0.020
C ₁₀	0.0016	0.0040	0.0017	0.0020	0.0021	0.0014		
C ₁₁	0.0018	0.0039	0.0017	0.0023	0.0020	0.0017		
C ₁₂	0.0018	0.0034	0.0015	0.0021	0.0020	0.0014		
C ₁₃	0.0017	0.0034	0.0018	0.0020	0.0020	0.0014		
N.....	0.0012	0.0033	0.0013	0.0019	0.0016	0.0011		
O.....	0.0014	0.0031	0.0013	0.0013	0.0014	0.0012		
Cl.....	0.0004	0.0008	0.0004	0.0003	0.0006	0.0003		
Room temperature								
C ₁	0.0014	0.0028	0.0015	0.0029	0.0037	0.0027	H ₃	0.010
C ₂	0.0014	0.0026	0.0016	0.0031	0.0038	0.0028	H ₄	0.012
C ₃	0.0016	0.0031	0.0020	0.0034	0.0046	0.0033	H ₆	0.007
C ₄	0.0018	0.0030	0.0017	0.0038	0.0041	0.0032	H ₇	0.016
C ₅	0.0017	0.0027	0.0016	0.0033	0.0037	0.0026	H ₉	0.015
C ₆	0.0015	0.0026	0.0016	0.0030	0.0037	0.0029	H ₁₀	0.014
C ₇	0.0015	0.0027	0.0018	0.0031	0.0035	0.0029	H ₁₁	0.021
C ₈	0.0017	0.0030	0.0017	0.0034	0.0039	0.0028	H ₁₂	0.010
C ₉	0.0016	0.0025	0.0021	0.0034	0.0043	0.0030	H ₁₃	0.016
C ₁₀	0.0020	0.0031	0.0022	0.0041	0.0045	0.0037		
C ₁₁	0.0024	0.0029	0.0019	0.0044	0.0041	0.0035		
C ₁₂	0.0025	0.0033	0.0019	0.0046	0.0045	0.0035		
C ₁₃	0.0019	0.0026	0.0019	0.0041	0.0041	0.0034		
N.....	0.0011	0.0021	0.0013	0.0026	0.0032	0.0022		
O.....	0.0012	0.0026	0.0016	0.0027	0.0031	0.0024		
Cl.....	0.0004	0.0007	0.0003	0.0009	0.0011	0.0009		

TABLE 6.

C-H bond lengths and angles.

Bond	Length (Å)	Low temperature			
		Angle		Angle	
C ₃ -H ₃	1.00	C ₂ -C ₃ -H ₃	118.2°	C ₄ -C ₃ -H ₃	121.5°
C ₄ -H ₄	1.07	C ₃ -C ₄ -H ₄	125.3	C ₅ -C ₄ -H ₄	115.3
C ₅ -H ₆	0.97	C ₅ -C ₆ -H ₆	122.1	C ₇ -C ₆ -H ₆	117.8
C ₇ -H ₇	0.99	C ₇ -C ₇ -H ₇	118.9	N-C ₇ -H ₇	119.6
C ₉ -H ₉	1.08	C ₈ -C ₉ -H ₉	121.6	C ₁₀ -C ₉ -H ₉	118.3
C ₁₀ -H ₁₀	0.99	C ₉ -C ₁₀ -H ₁₀	121.9	C ₁₁ -C ₁₀ -H ₁₀	116.0
C ₁₁ -H ₁₁	1.08	C ₁₀ -C ₁₁ -H ₁₁	121.2	C ₁₂ -C ₁₁ -H ₁₁	120.3
C ₁₂ -H ₁₂	1.03	C ₁₁ -C ₁₂ -H ₁₂	116.6	C ₁₃ -C ₁₂ -H ₁₂	123.3
C ₁₃ -H ₁₃	0.99	C ₁₂ -C ₁₃ -H ₁₃	113.8	C ₈ -C ₁₃ -H ₁₃	124.4
Room temperature					
C ₃ -H ₃	1.03	C ₂ -C ₃ -H ₃	114.9	C ₄ -C ₃ -H ₃	124.2
C ₄ -H ₄	1.05	C ₃ -C ₄ -H ₄	120.9	C ₅ -C ₄ -H ₄	118.7
C ₅ -H ₆	0.97	C ₅ -C ₆ -H ₆	122.8	C ₇ -C ₆ -H ₆	116.5
C ₇ -H ₇	0.93	C ₇ -C ₇ -H ₇	116.9	N-C ₇ -H ₇	121.3
C ₉ -H ₉	1.08	C ₈ -C ₉ -H ₉	117.7	C ₁₀ -C ₉ -H ₉	122.2
C ₁₀ -H ₁₀	0.94	C ₉ -C ₁₀ -H ₁₀	122.1	C ₁₁ -C ₁₀ -H ₁₀	116.7
C ₁₁ -H ₁₁	1.00	C ₁₀ -C ₁₁ -H ₁₁	117.7	C ₁₂ -C ₁₁ -H ₁₁	122.3
C ₁₂ -H ₁₂	0.91	C ₁₁ -C ₁₂ -H ₁₂	120.8	C ₁₃ -C ₁₂ -H ₁₂	118.1
C ₁₃ -H ₁₃	1.03	C ₁₂ -C ₁₃ -H ₁₃	112.3	C ₈ -C ₁₃ -H ₁₃	126.6

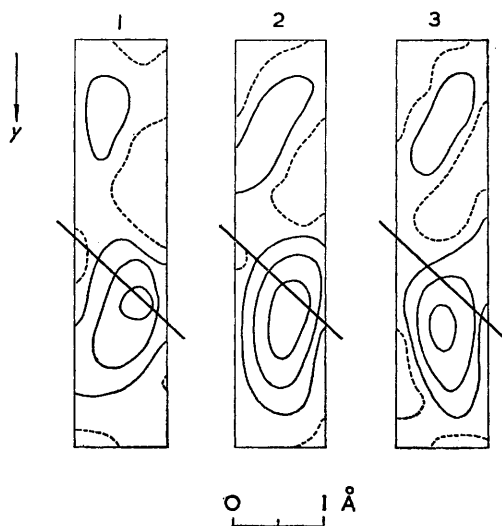


FIG. 3. *N*-5-Chlorosalicylideneaniline; parallel sections of $\delta(xyz)$ in the $O \cdots N$ region at low temperature. Contour interval $0.125 \text{ e}/\text{\AA}^3$; zero contour dotted. The co-ordinates xyz of the upper left corners are: 1 (0.323, 0.134, 0.203); 2 (0.342, 0.134, 0.206); 3 (0.360, 0.134, 0.209). Distance between sections 0.2 \AA . The trace of ring *c* is shown.

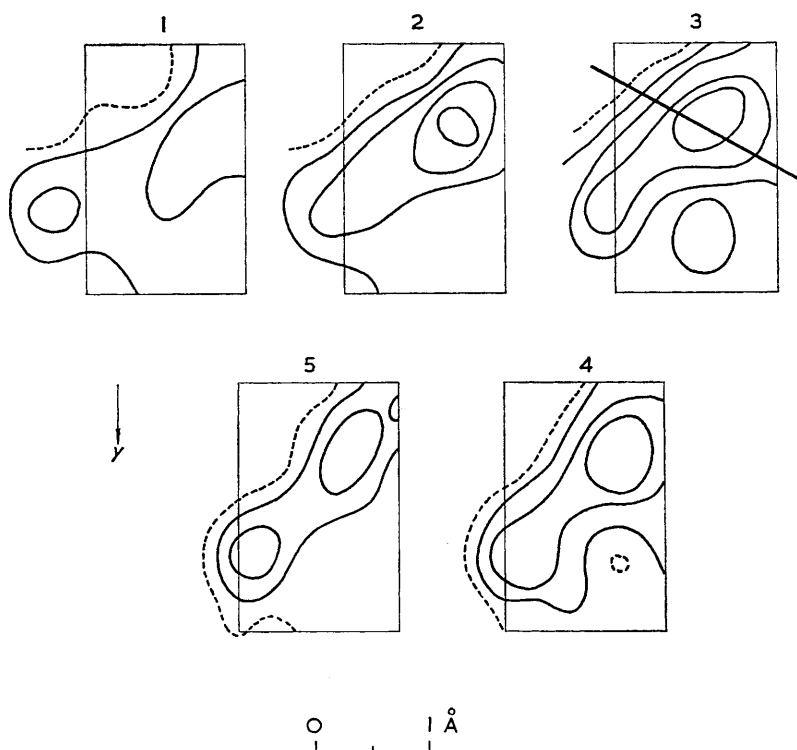


FIG. 4. *N*-5-Chlorosalicylideneaniline; parallel sections of $\delta(xyz)$ in the $O \cdots N$ region at room temperature. Contour interval $0.063 \text{ e}/\text{\AA}^3$; zero contour dotted. The co-ordinates (xyz) of the upper left corners are 1 (0.305, 0.486, 0.179); 2 (0.293, 0.486, 0.185); 3 (0.281, 0.486, 0.192); 4 (0.267, 0.486, 0.198); 5 (0.255, 0.486, 0.204). Distance between sections 0.2 \AA . The trace of ring *c* is shown in section 3 only.

TABLE 7.

(a) Equations of best planes.

Low temperature					Room temperature				
					m_1	m_2	m_3	d	
Ring A	C ₁ ...C ₆	2.4879	3.3533	-12.1438	0.0711	2.3929	3.4110	-12.2357	0.1116
Ring B	C ₈ ...C ₁₃	2.5057	3.2929	-12.4365	0.0576	2.4445	3.3608	-12.4655	0.1346
Ring c	O, C ₂ , C ₁ , C ₇ , N	2.4601	3.2504	-12.6676	-0.0703	2.3408	3.3246	-12.6934	-0.0109

Equation of best plane: $m_1x + m_2y + m_3z - d = 0$, where xyz are fractional co-ordinates.

(b) Deviations (Å) from best planes.

Atom	Low temperature		Room temperature		Atom	Low temperature	Room temperature
	Ring A	Ring c	Ring A	Ring c		Ring B	Ring B
C ₁	+0.007	+0.013	+0.008	+0.008	C ₈	-0.009	-0.012
C ₂	-0.004	+0.007	+0.006	+0.008	C ₉	+0.007	+0.013
C ₃	-0.002		-0.016		C ₁₀	-0.002	-0.006
C ₄	+0.004		+0.011		C ₁₁	-0.001	-0.001
C ₅	+0.000		+0.004		C ₁₂	-0.002	+0.002
C ₆	-0.006		-0.013		C ₁₃	+0.007	+0.005
C ₇	+0.013	-0.028	+0.018	-0.021	N	-0.005	-0.022
O	+0.015	-0.013	+0.023	-0.011			
Cl	+0.018		+0.012				
N	+0.100	+0.021	+0.088	+0.016			

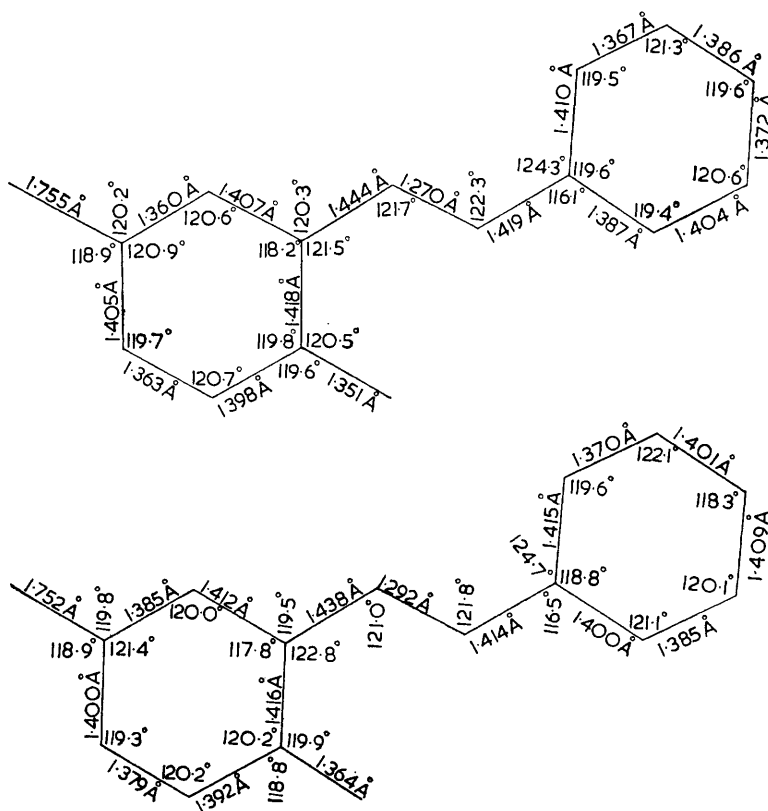
FIG. 5. *N*-5-Chlorosalicylideneaniline; experimentally determined bond lengths and bond angles at room temperature (top) and near 90°K (bottom).

TABLE 8.

Structure factors.

Low temperature data																							
<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c			
0	0	4	2421	-1366	2300	2675	8	0	6	4001	-3303	2490	4136	1	1	15	955	-113	-811	819			
		6	4927	3665	3281	4919			7	5454	-4073	-4577	5978			16	1952	1295	-1491	1974			
		8	3176	3046	-852	3163			8	4017	-1085	-4096	4237			17	926	127	861	871			
		10	5454	5392	-34	5393			9	2648	1437	-2180	2611			18	2535	-2032	-1514	2534			
		12	2933	2516	-1361	2860			10	1751	-636	1780	1891			19	1197	596	-1141	1287			
		14	4046	3553	-1506	3859			11	1880	952	-1888	2114			20	854	-387	-760	853			
		16	1952	1886	-854	2071			12	3832	-3324	2342	4066			21	984	-540	-679	868			
		18	4088	4209	975	4320			13	3561	3555	-1231	3743			22	1240	1204	146	1213			
		20	7277	6852	3942	7905			14	3205	-1067	-3168	3342			23	783	-685	98	692			
		22	3832	-2804	-2119	3514			15	2392	-1722	-1851	2528			24	1823	1943	623	2040			
		24	1253	1119	-349	1172			16	2308	-2211	1065	2454			2	1	0	853	800	0	800	
2	0	0	6024	-6651	0	6651			17	1553	-1186	-863	1466			2	1	0	4785	-1518	4871	5102	
		1	5739	-6174	2140	6534			18	1026	-1020	0	1020			2	1894	1351	1405	1949			
		2	5583	4003	-4342	5906			19	770	-453	-656	797			3	6865	5828	5288	7870			
		3*	8176	-1542	9384	9509			20	1467	-1529	-259	1551			4	5341	2379	-5069	5600			
		4	4017	3872	448	3898			21	1438	-104	-1545	1549			5	5447	-1840	2067	3491			
		5	3191	-1276	3028	3286			10	0	512	464	484			6	4615	4241	-1961	4673			
		6	5014	4378	-1861	4757			1	3574	429	-3895	3919			7	4286	-3683	2439	4418			
		7	5924	1461	5696	5784			2	770	755	206	753			8	3859	2382	2952	3793			
		8	3645	2815	-2112	3519			3	2635	-1099	-2480	2713			9	3817	1758	3150	3607			
		9	1936	1395	774	1595			4	2337	256	-2566	2579			10	3347	3337	-33	3337			
		10	5198	4886	1604	5143			5	3004	1855	-2637	3213			11	2635	-958	2675	2688			
		11	1809	-238	1739	1755			6	256	93	287	301			12	2221	1964	-692	2083			
		12	2492	-432	1963	2010			7	3289	2812	-2000	3451			13	4117	2396	3353	4122			
		13	4457	-785	4359	4429			8	3334	1906	3078	3621			14	2078	771	2026	2168			
		14	2279	2179	300	2199			9	2877	-2643	-1063	2848			15	1923	-600	1947	2038			
		15	2463	743	2347	2461			10	1509	-585	-1497	1607			16	398	257	219	337			
		16	1482	1363	-391	1418			11	1224	459	-1218	1302			17	2435	-1808	1663	2457			
		17	3305	22	3549	3549			12	3091	2873	-1340	3170			18	2350	1689	-1553	2298			
		18	<200	3	-33	33			13	4428	-2634	-4069	4847			19	2535	868	2453	2602			
		19	1696	1133	1255	1691			14	1424	-1244	764	1460			20	868	878	-30	878			
		20	1253	373	-1312	1364			15	2830	418	-2564	2598			21	2078	-124	1985	1989			
		21	2308	-1614	971	1884			16	1294	-395	1003	1078			22	1625	1398	-297	1429			
		22	3803	3468	-105	3470			17	2250	1213	-1714	2099			23	2337	841	2061	2226			
		23	2835	-456	2650	2689			18	1110	906	-560	1065			24	1722	1845	-579	1933			
		24	1026	891	404	979			12	0	912	815	0			3	1	0	171	87	0	87	
4	0	0	4302	-4213	0	4213			1	570	241	-564	613			3	1	0	3489	-613	-1687	-161	3690
		1	4842	1145	4799	4934			2	3075	3309	-901	3429			2	2292	2126	-895	2307			
		2	5755	-939	-5840	5915			3	1965	-1989	-648	2092			3	3062	3074	-1037	3244			
		3	5454	-2858	5039	5793			4	2577	2133	1956	2894			4	2179	-268	-2019	2037			
		4	4357	-2122	3537	4125			5	1197	428	-1189	1264			5	1153	1112	-94	1116			
		5	3532	2533	2616	3642			6	2677	2619	-682	2707			6	4215	2755	2680	3844			
		7	4215	-1029	4240	4363			7	1197	856	-867	1218			7	741	-54	-728	730			
		8	1823	-328	1742	1772			8	984	986	-197	1006			8	1068	-959	243	989			
		9	4814	319	4873	4883			9	1153	-524	-1088	1208			9	2392	-1320	1844	2267			
		10	4247	-7677	-2426	8051			10	1809	1788	-95	1790			10	699	-68	-666	669			
		11	3503	-782	3449	3536			11	1068	694	-692	980			11	1097	492	743	891			
		12	2848	2704	304	2721			12	1580	1457	-132	1463			12	1295	945	-680	1164			
		13	2036	224	2068	2080			13	1282	-345	-1220	1268			13	1211	121	-1020	1028			
		14	826	-759	504	911			14	2877	2733	-90	2734			14	2065	1974	-160	1981			
		15	1509	-390	1442	1493			15	599	405	-413	579			15	854	-533	-726	901			
		16	1068	-993	-436	1084			14	0	1936	1870	0			16	1268	-762	1012	1267			
		17	3162	480	3224	3259			1	841	-564	635	849			17	456	-216	-252	332			
		18	2379	-1896	1540	2443			2	1794	1278	871	1546			18	1240	1247	110	1252			
		19	2107	-366	2091	2123			3	1894	861	1457	1892			19	1055	839	-815	1170			
		20	1667	-824	-1428	1649			4	2906	2791	-488	2834			20	427	116	-381	398			
		21	2036	360	1850	1885			5	3162	-10	2904	2904			21	1010	-537	-743	917			
		22	456	-204	-349	404			6	227	163	44	168			22	1298	596	60	599			
		23	1909	-588	1610	1714			7	1438	-1173	-446	1255			23	623	1087	408	1161			
		6	0	0	0	0			8	1923	1982	106	1985			4	1	0	2107	2072	0	2072	
		1	4101	-3882	-1601	4199			9	955	341	865	930			4	1	0	4885	236	5139	5145	
		2	6367	-5081	3716	6295			10	2023	2056	178	2064			2	3803	-3650	988	3781			
		3	3975	3516	1619	3870			0	1	2	6380	7406			3	3120	-298	3097	3111			
		4	3690	-3526	248	3535			4*	10,980	9307	13,024	16,007			4	1696	-494	-1695	1766			
		5	1923	325	1814	1843			6	5383	2975	-4348	5268			5	4331	674	4245	4298			
		6	2848	-2713	-848	2842			8	2194	2236	-422	2276			6	4671	-2629	3214	4153			
		7	314	249	229	338			10	1936	1567	-821	1769			7	3405	-510	3233	3273			
		8	4030	-4079	-29	4079			12	2906	3018	-377	3042			8	2706	-2006	-1825	2712			
		9	1395	-687	1181	1319			14	3432	3383	-119	3385			9	3191	395	3207	3231			
		10	3091	-3041	-291	3055			16	6523	6297	-1999	6606			10	841	-86	690	695			
		11	868	-425	774	883			18	2221	1195	2152	2462			11	2719	-607	2612	2682			
		12	3460	-3526	-57	3527			20	1809	1673	-331	1705			12	1880	-2036	54	2036			
		13	398	-236	299	381			22	2308	2086	1003	2315			13	2848	27	2887	2887			
		14	1952	-1878	712	2009			24	1696	1066	1578	1904			14	1767	-391	-1757	1781			
		15	456	-170	449	480			1	1	2933	-1497	-2682			15	2463	-57	2553	2354			
		16	2577	-2480	-853	2623			2	2906	-2779	-641	2852			16	2052	-1522	1498	2135			
		17	754	-751	-5	751			3	3716	-2538	3330	4108			17	1767	847	1906	1937			
		18																					

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TABLE 8. (Continued.)

100kF _o				100A _c				100B _c				100F _c			
<i>h</i>	<i>k</i>	<i>l</i>		<i>h</i>	<i>k</i>	<i>l</i>		<i>h</i>	<i>k</i>	<i>l</i>		<i>h</i>	<i>k</i>	<i>l</i>	
5	1	5	2363	984	1974	2206		9	1	8	2550	-1836	1819	2584	
		6	2933	-1891	-2109	2833				9	1353	143	1430	1437	
		7	1467	-1325	-738	1517				10	854	667	-623	913	
		8	1767	1574	782	1758				11	1838	-1911	209	1923	
		9	712	568	394	691				12	1880	-790	-2004	2155	
		10	485	351	-44	354				13	1126	-649	-972	1169	
		11	498	-485	281	561				14	1026	-148	1030	1040	
		12	770	702	167	722				15	343	-259	180	315	
		13	1211	1119	416	1194				16	1651	-982	1040	1431	
		14	1395	-1455	33	1456				17	1923	1746	-127	1750	
		15	1538	-1188	1130	1640				18	1110	-32	-1054	1054	
		16	1738	1610	-579	1711				19	470	-474	-44	477	
		17	699	340	548	645		10	1	0	1865	-2024	0	2024	
		18	1211	1292	-318	1331				1	2693	293	-2708	2724	
		19	541	-116	-806	617				2	754	746	108	754	
		20	1039	-546	771	945				3	4486	1607	-4565	4840	
		21	754	421	481	639				4	2392	2484	-82	2485	
		22	1311	20	1313	1313				5	541	319	-502	595	
		23	1068	-900	815	1214				6	427	-284	425	511	
6	1	0	3476	-3305	0	3305				7	2777	171	-2875	2880	
		1	754	541	-418	684				8	1809	991	-1572	1858	
		2	2791	-2621	-963	2792				9	1253	-643	-887	1096	
		3	1625	250	1569	1589				10	527	163	452	480	
		4	3432	-3299	768	3387				11	1722	-262	-1712	1732	
		5	1738	326	-1547	1581				12	1010	-54	1175	1176	
		6	5512	-5387	-737	5437				13	2791	-352	-3046	3066	
		7	3247	1259	3041	3291				14	1197	942	-691	1168	
		8	3532	-3055	835	3167				15	1909	-590	-1779	1874	
		9	1010	-533	897	1044				16	1567	1237	581	1367	
		10	1126	-970	152	982				17	3803	380	-3695	3714	
		11	556	-379	518	642				18	955	-443	-797	912	
		12	3574	-3585	-495	3619		11	1	0	456	-376	0	376	
		13	1467	-1425	461	1498				1	<227	13	105	106	
		14	3191	-3355	-211	3362				2	485	-495	2	495	
		15	1168	889	-832	1217				3	912	518	-693	865	
		16	1495	-1495	85	1497				4	1809	796	1710	1886	
		17	984	-291	995	1037				5	1240	-1193	35	1194	
		18	2094	-2110	639	2205				6	1282	-1070	-925	1415	
		19	398	454	122	470				7	754	145	-783	796	
		20	2036	-2010	-274	2028				8	441	-379	-204	431	
		21	456	166	428	459				9	485	100	-503	512	
		22	1538	-1584	119	1588				10	955	-1006	-3	1006	
7	1	0	1068	-962	0	962				11	583	255	-611	662	
		1	1667	21	1610	1611				12	<185	-53	81	105	
		2	556	-162	-463	491				13	485	380	-265	464	
		3	683	435	497	660				14	683	-609	-220	648	
		4	570	89	-554	561				15	612	262	499	564	
		5	1424	-1210	-378	1268				16	1340	1237	-558	1367	
		6	1224	-1014	316	1062				17	641	-448	-606	754	
		7	628	160	459	467		12	1	0	4386	4803	0	4803	
		8	1010	621	-771	990				1	441	395	218	451	
		9	1667	-1651	-28	1651				2	1224	1234	504	1333	
		10	995	516	773	929				3	1424	-95	-1452	1452	
		11	3062	2741	1284	3027				4	1139	1210	-35	1210	
		12	812	-391	611	726				5	1224	-709	-936	1174	
		13	670	-609	192	639				6	2450	2618	310	2656	
		14	1010	-745	-693	1018				7	1340	12	-1528	1528	
		15	541	470	324	571				8	1240	1360	-444	1431	
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		19	826	-80	753	757				12	1153	1145	-21	1145	
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		21	641	-133	602	617				14	2250	2284	-263	2299	
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		7	1809	-298	-1718	1744				6	1095	132	-986	997	
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		9	4529	1822	-4447	4734				8	968	-208	907	930	
		10	4814	-4976	-1354	5157				9	1809	-116	-1657	1661	
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		15	854	474	-771	905		14	1	0	2107	1978	0	1978	
		16	2435	-2547	-307	2565				1	1424	670	1144	1326	
		17	712	-51	-647	649				2	1952	1748	438	1802	
		18	385	-350	-25	351				3	1651	-549	1478	1574	
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		20	1965	-1977	-282	1997				5	414	264	281	385	
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		4	1909	45	-1972	1972				10	1010	1069	-201	1088	
		5	1851	1637	975	1905		15	1	0	1282	1251	0	1251	
		6	527	-157	452	479				1	470	436	8	436	
		7	1197	1190	66	1192				2	1068	-801	310	859	
15	1	3	456	-76	-431	438			</						

TABLE 8. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	100 <i>k</i> F _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>k</i> F _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>k</i> F _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c		
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		13	2337	-326	2069	2094			18	1838	-1623	-371	1665			16	1366	1212	-320	1253		
		14	1168	-1274	272	1303			19	1139	-261	-1011	1044			18	2236	2222	159	2228		
		15	3859	833	3807	3897	9	2	0	3276	-3077	0	3077			20	1424	1422	-403	1478		
		16	897	-3	-853	853			1	2036	1942	43	1943		1	3	0	3289	-3506	0	3506	
		17	1667	-455	1479	1947			2	783	-152	-683	700		1	2	0	2107	-1163	-1924	2248	
		18	498	447	-135	467			3	826	473	-726	866		2	1	0	1580	1410	-478	1489	
		19	1823	46	1472	1473			4	1324	-486	1223	1316		3	3	0	628	-41	-697	698	
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		22	1366	-647	1173	1339			7	498	-297	355	462		6	1	0	1153	1051	-618	1219	
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		17	1240	695	-836	1087			10	654	-543	-30	543		2	1	0	1936	1828	981	2074	
		18	2991	-2504	-125	2507			11	955	708	-338	784		3	1	0	583	334	-465	573	
		19	1880	-267	1669	1690			12	485	-436	-265	510		4	1	0	1324	1350	112	1355	
		20	1823	-1764	-219	1777			13	812	-187	-725	748		5	1	0	683	357	-456	579	
		21	272	45	291	294			14	912	-721	-461	856		6	1	0	1794	1751	447	1807	
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		4	1359	-1042	-722	1267			3	1253	-140	-1110	1110		11	1	0	579	-232	153	278	
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		9	683	107	-6																	

[1964]

Topochemistry. Part IX.

2079

TABLE 8. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	100kF _o	100A _c	100B _c	100F _c	<i>h</i>	<i>k</i>	<i>l</i>	100kF _o	100A _c	100B _c	100F _c	<i>h</i>	<i>k</i>	<i>l</i>	100kF _o	100A _c	100B _c	100F _c
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3			1285	786	909	1202	9			926	226	931	958	1			1680	-1080	-1462	1806
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5			997	-396	908	990	11			1253	534	1231	1342	3			2350	-746	-2589	2694
6			1126	893	433	992	12			628	-488	483	687	4			1553	-1118	1135	1593
7			1353	129	1355	1361	13			1126	761	1005	1260	5			1110	957	-628	1145
8			841	847	230	877	14			1382	-615	-1139	1294	6			783	498	-697	857
9			1823	-562	1546	1645	15			1896	-188	1558	1569	7			2435	-72	-2601	2602
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15			556	69	631	634	2			1366	-796	1095	1355	13			2078	342	-1888	1919
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3			3318	-1513	3069	3422	10			1567	-1587	-190	1599	7			1197	63	-1101	1103
4			1738	-1683	-20	1683	11			797	128	-829	839	8			1709	-1506	-949	1780
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6			741	-662	262	712	13			2007	-145	-2016	2021	10			498	-457	196	497
7			854	343	762	835	14			1851	-1998	396	2037	11			1110	341	-957	1016
8			2392	-2327	-250	2341	15			1709	-458	-1575	1635	12			1282	-1466	282	1492
9			699	-382	-526	650	16			725	-195	-746	771	12	3	0	725	849	0	849
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5			1268	-603	1069	1227	13			1580	-1531	-309	1562	5			868	155	-1079	1090
6			1609	-1084	1032	1496	14			1638	-1465	615	1589							
			854	-583	588	828	15			512	-28	509	509							

Room temperature data

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16			841	830	-238	864	10			1987	-1972	-139	1977	7			2228	1908	-1190	2249
17			2154	293	2113	2133	11			736	-197	631	661	8			2375	1513	1789	2343
18			284	37	256	259	12			2407	-2334	-19	2334	9			1576	-1498	82	1499
19			683	299	621	689	13			410	-267	218	345	10			1166	-871	-861	1225
20			442	-16	-507	507	14			989	-922	341	983	11			652	316	-521	609
21			1335	-1104	723	1320	15			220	-32	240	242	12			1808	1624		

TABLE 8. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	
12	0	2	1849	1836	-640	1945	3	1	4	1713	-439	-1557	1617	1	1	1283	-105	1252	1257		
		3	998	-989	-324	1041			5	925	843	-172	860			2	568	-231	-459	514	
		4	1576	1023	1236	1605			6	3311	2363	2290	3291			3	200	213	56	220	
		5	516	186	-494	528			7	483	-76	-467	474			4	568	235	-432	492	
		6	1135	1118	-431	1198			8	1450	-1328	337	1370			5	1376	-1174	-542	1393	
		7	631	462	-473	661			9	2197	-934	1883	2101			6	893	-793	213	821	
		8	232	195	-61	204			10	504	-263	-354	441			7	220	121	96	154	
		9	525	-292	-460	545			11	810	568	471	738			8	640	356	-494	609	
		10	967	868	27	868			12	640	411	-471	625			9	1166	-1122	-113	1127	
		11	640	381	-445	586			13	788	-105	-685	693			10	609	285	548	617	
		12	463	426	-190	467			14	1209	1167	-253	1194			11	1944	1928	256	1945	
		13	410	-115	-360	377			15	568	-480	-358	598			12	389	-187	296	350	
		14	1094	1055	-142	1064			16	956	-375	924	997			13	578	-578	84	584	
		15	325	321	-68	329			17	241	-97	-100	139			14	693	-586	-334	674	
		16	568	575	-212	613			18	547	542	-115	554			15	<241	237	62	245	
14	0	0	956	909	0	909			19	662	587	307	662			16	232	-37	245	248	
		1	315	-234	153	280			20	<220	-98	-78	125			17	631	-464	602	684	
		2	547	212	445	493			21	284	240	-134	275			18	<210	-64	-29	70	
		3	862	615	442	757			22	<220	177	140	226			19	<232	-22	255	256	
		4	1376	1238	-334	1283			23	483	461	42	463			20	315	-227	-224	319	
		5	1450	114	1332	1336			24	126	64	147	160			21	<168	-187	128	227	
		6	315	-79	243	255			4	1	0	2176	0	2176		22	<115	-7	-78	78	
		7	989	-853	-278	898			1	4309	331	4500	4512		8	1	0	2627	-2565	0	2565
		8	662	637	-90	644			2	3268	-8099	975	3249			1	1576	-396	-1437	1491	
		9	315	278	197	341			3	2816	-419	2914	2944			2	640	-512	165	538	
		10	736	708	85	713			4	1608	-833	-1398	1627			3	1240	-1069	-427	1151	
		11	337	187	268	327			5	3373	540	3430	3472			4	2701	-2748	-398	2777	
0	1	2	6989	6830	-2344	7221			6	3531	-1919	2868	3451			5	1913	-890	-1567	1802	
		3	*10,552	9771	11,890	15,390			7	3026	-464	2860	2898			6	368	-211	-263	338	
		4	4309	2197	-3764	4358			8	2427	-1782	-1534	2351			7	1271	-251	-1254	1279	
		5	1691	1695	-351	1731			9	2427	262	2406	2421			8	1061	-426	852	953	
		6	1481	1412	-168	1422			10	547	19	473	473			9	2859	824	-2802	2921	
		7	2281	2321	-374	2351			11	2207	-355	2024	2054			10	3394	-3389	-616	3444	
		8	2701	2617	-284	2632			12	1481	-1349	313	1385			11	1567	-406	-1544	1596	
		9	3741	3392	-1636	3766			13	1913	-46	1872	1873			12	673	518	-453	688	
		10	1796	893	1652	1878			14	1440	-737	-1219	1425			13	494	541	20	541	
		11	841	827	-201	851			15	1586	-87	1512	1515			14	819	-781	356	858	
		12	1146	1046	266	1079			16	1113	-563	948	1103			15	516	185	-493	527	
		13	946	736	654	985			17	883	160	947	960			16	1199	-1320	39	1321	
1	1	0	5278	-6416	0	6416			18	357	-383	-127	403			17	210	-40	-280	283	
		1	3069	-1678	-2343	2882			19	1177	-151	1285	1294			18	273	-41	-309	312	
		2	3605	-3348	-382	3369			20	210	42	172	177			19	915	-411	-789	890	
		3	4068	-2447	-3522	4288			21	810	86	749	754			20	736	-748	-1	748	
		4	7820	8086	-3353	8754			22	651	-576	87	583			21	535	-352	-450	571	
		5	2565	1661	-1881	2509			23	525	-47	528	531		9	1	0	819	-803	0	803
		6	1240	-492	1046	1156			5	1	0	568	-512	0	512	1	0	<232	-18	-207	207
		7	714	-156	-708	725			1	1703	1409	-1007	1732			2	494	-293	376	477	
		8	946	662	-537	847			2	3721	1923	3143	3655			3	1251	-964	-790	1246	
		9	967	190	-931	950			3	2269	-1087	1953	2253			4	1415	-59	-1466	1467	
		10	778	-247	631	678			4	494	-209	438	465			5	1251	1244	367	1297	
		11	200	62	-159	171			5	1755	1076	1366	1739			6	378	222	373	434	
		12	819	296	736	793			6	2785	-1978	-1606	2548			7	714	-712	-154	729	
		13	220	201	-195	280			7	1567	-1384	-610	1513			8	1493	-931	1158	1486	
		14	325	298	213	366			8	1376	1209	518	1315			9	1051	258	1020	1052	
		15	294	64	-261	369			9	389	333	137	360			10	778	545	-609	818	
		16	1041	775	661	1019			10	220	92	-3	92			11	1314	-1299	228	1319	
		17	873	378	551	668			11	420	-380	166	415			12	1187	-631	-1044	1220	
		18	1586	-1563	-422	1619			12	399	410	44	413			13	841	-502	-678	844	
		19	588	120	-578	590			13	736	705	65	708			14	535	217	512	556	
		20	273	-212	-289	359			14	1051	-986	317	1035			15	<220	-99	-26	103	
		21	337	-281	-127	308			15	998	-689	785	1030			16	704	-340	540	638	
		22	599	576	-157	597			16	831	681	-512	852			17	925	773	-390	866	
		23	200	-119	101	156			17	<241	154	80	174			18	599	-90	-563	570	
		24	778	781	-4	781			18	463	461	-87	469			19	168	-221	16	222	
		25	1082	1012	0	1012			19	410	-301	-348	460			20	126	-127	-11	127	
2	1	0	4572	-1545	4615	4867			20	640	-148	540	560		10	1	0	1104	-1248	0	1248
		1	1954	1202	1555	1966			21	<220	250	150	292			1	1555	-1	-1594	1594	
		2	6884	5465	4697	7206			22	452	73	443	449			2	389	421	-27	422	
		3	4908	1966	-4636	5036			23	473	-236	410	473			3	2838	942	-2819	2972	
		4	2995</																		

[1964]

Topochemistry. Part IX.

2081

TABLE 8. (Continued.)

[Left Group]				[Right Group]									
<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c
11	1	8	<241	-51	-122	132	2	2	0	1345	1331	0	1331
		9	368	91	-387	397			1	2154	1174	1565	1957
		10	420	-406	-1	406			2	1524	1277	-672	1443
		11	442	141	-464	485			3	4141	-1384	4013	4245
		12	<210	189	-21	190			4	2880	2754	-765	2859
		13	<241	236	-7	236			5	1471	1259	248	1283
		14	263	-273	-100	290			6	494	-445	132	465
		15	389	217	321	387			7	4844	259	5030	5037
		16	726	616	-391	729			8	2764	2898	-288	2913
		17	253	-213	-207	297			9	578	-168	531	557
12	1	0	2480	2613	0	2613			10	1839	1787	382	1827
		1	241	324	114	343			11	1376	96	1433	1437
		2	652	476	481	677			12	1113	1016	275	1052
		3	673	-123	-653	664			13	2480	-442	2405	2445
		4	410	352	-35	354			14	358	-189	-169	253
		5	547	-361	-382	525			15	<253	-9	164	164
		6	1240	1302	42	1303			16	1230	1195	396	1259
		7	693	1	-698	698			17	1534	421	1467	1526
		8	442	406	-169	439			18	452	380	-124	400
		9	294	130	-257	288			19	389	-35	415	417
		10	1292	1184	-46	1185			20	621	494	-271	563
		11	273	133	-274	304			21	736	107	717	725
		12	337	359	66	365			22	516	494	-57	497
		13	557	294	-426	517			23	621	-7	647	647
		14	851	725	-143	739			3	2	253	-284	0
		15	284	279	-57	285			1	578	-246	-477	537
13	1	0	346	-376	0	376			2	2637	2483	-553	2544
		1	463	413	-342	536			3	557	-600	43	601
		2	430	450	161	478			4	2081	-1024	1650	1942
		3	368	316	199	373			5	915	-213	-899	924
		4	210	-258	-86	272			6	1261	652	1101	1280
		5	<210	-79	-43	90			7	3017	2588	1343	2915
		6	516	179	-487	519			8	1787	761	-1566	1741
		7	504	-414	181	452			9	430	19	360	361
		8	483	4	404	404			10	452	16	460	461
		9	714	-154	-684	702			11	1187	-1201	101	1205
		10	714	706	-67	709			12	1051	649	805	1034
		11	798	724	-182	746			13	1240	-1179	130	1179
		12	399	-345	-134	370			14	1094	331	-949	1005
		13	442	-5	454	454			15	1156	1100	-373	1161
14	1	0	683	636	0	636			16	<253	1	156	156
		1	473	369	297	474			17	<241	172	134	218
		2	788	685	118	695			18	1082	935	619	1121
		3	736	-323	664	739			19	<220	35	-212	215
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		9	315	-194	180	265			2	2439	837	2173	2329
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		1	<168	202	119	234			5	3815	-1889	3491	3969
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		6	<126	63	-9	63			10	1113	-998	139	1007
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		6	4194	4353	18	4353			13	1461	-208	1410	1426
		8	2448	-373	2519	2546			14	631	-603	167	625
		10	3332	3136	-1201	3358			15	2092	731	2048	2174
		12	2007	1078	-1565	1900			16	504	-229	-455	509
		14	2859	2747	677	2829			17	746	-362	725	811
		16	1156	1068	193	1085			18	284	184	-185	261
		18	1030	1077	-99	1082			19	746	-117	692	702
		20	1094	976	14	976			20	1041	-689	634	936
		22	578	585	-26	585			21	714	28	685	686
1	2	0	253	260	0	260			22	399	-144	335	365
		1	693	524	-273	591			5	2	631	633	0
		2	420	-81	-364	373			1	1503	308	1375	1409
		3	2618	2507	-680	2598			2	1187	413	1017	1098
		4	3583	-634	-3467	3524			3	1166	-1028	-365	1091
		5	1691	-1297	-817	1533			4	1113	501	-941	1066
		6	893	-211	-788	815			5	946	779	504	927
		7	2386	-2275	62	2276			6	1503	-743	1337	1530
		8	4161	3347	2708	4305			7	1051	-503	925	1054
		9	1902	1319	-1370	1902			8	516	324	-395	511
		10	463	167	-304	347			9	442	185	387	429
		11	1020	362	-900	970			10	378	327	-76	335
		12	1356	1268	-450	1339			11	609	-454	386	596
		13	1471	1334	521	1432			12	253	-225	188	293
		14	1913	-1232	1626	2040			13	253	-100	226	248
		15	599	-367	-443	575			14	253	162	-304	345
		16	621	-450	353	560			15	253	-318	124	341
		17	621	-470	-342	581			16	652	547	479	727
		18	621	129	-557	572			17	442	347	241	422
		19	284	306	-38	308			18	<220	38	-120	126
		20	420	293	-176	342			19	599	-318	455	555
		21	358	281	-190	339			20	378	21	347	348
		22	232	41	-165	170			21	<179	-84	49	98
		23	263	271	-111	293			22	<126	-38	119	125
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		1	1555	-371	1434	1481			1	1555	-371	1434	1481
		2	3110	-2992	-248	3002			2	3110	-2992	-248	3002
		3	1755	710	-1466	1629			3	1755	710	-1466	1629
		4	1409	-1407	-28	1408			4	1409	-1407	-28	1408
		5	1082	618	892	1085			5	1082	618	892	1085
		6	1335	-1118	-947	1465			6	1335	-1118	-947	1465
		7	1314	-1278	405	1341			7	1314	-1278	405	1341
		8	2102	-1860	971	2098			8	2102	-1860	971	2098
		9	1481	1118	923	1450							

TABLE 8. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> ₀	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	
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			<220	125	-73	145				5	1545	567	1493	1597	12		1104	-1139	280	1173	12		1104	-1139	280	1173		
			13	819	-254	-646	694			6	1271	1288	231	1308	13		463	-44	446	448	13		463	-44	446	448		
			14	232	251	-65	259			7	831	730	549	913	14		399	-425	169	457	14		399	-425	169	457		
			15	652	22	-609	609			8	1902	1584	-1224	2002	15		483	-138	424	446	15		483	-138	424	446		
			16	325	-243	-26	244			9	2995	1288	3031	3293	16		1199	-1108	169	1121	16		1199	-1108	169	1121		
			17	673	-17	-668	669			10	1030	-987	317	1037	17		305	-14	257	257	17		305	-14	257	257		
11	2	0	2427	2546	0	2546	11			11	2806	-2502	1469	2901	18		977	-980	214	1003	18		977	-980	214	1003		
			1	253	-79	291	301			12	1892	1808	887	2014	19		578	-238	567	615	19		578	-238	567	615		
			2	746	-729	265	775			13	494	458	-258	526	7	3	0	504	-531	0	531	7	3	0	504	-531	0	531
			3	631	-498	-378	625			14	673	467	-569	736	8		1104	303	1094	1135	8		1104	303	1094	1135		
			4	851	-815	386	902			15	788	117	771	780	2		956	140	-946	957	2		956	140	-946	957		
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			6	<241	-93	-93	131			17	714	396	544	673	4		640	-603	155	623	4		640	-603	155	623		
			7	410	179	-397	436			18	410	381	106	396	5		494	-242	414	480	5		494	-242	414	480		
			8	<252	61	-130	144			19	767	219	708	741	6		946	-610	669	905	6		946	-610	669	905		
			9	516	-230	-400	461			20	557	503	113	516	7		368	-370	23	370	7		368	-370	23	370		
			10	<263	-63	15	65			21	463	-235	401	465	8		599	38	-576	577	8		599	38	-576	577		
			11	547	423	-201	468			3	3	0	1440	1348	0		<210	133	107	171	9		<210	133	107	171		
			12	325	-0	-257	257			1	767	654	206	686	10		210	-132	228	263	10		210	-132	228	263		
			13	662	25	-265	266			2	1283	653	961	1162	11		200	92	211	251	11		200	92	211	251		
			14	210	-229	-136	266			3	305	119	-196	230	12		273	-193	270	332	12		273	-193	270	332		
			15	368	334	-201	389			4	578	521	191	555	13		189	175	200	266	13		189	175	200	266		
12	2	0	746	806	0	806	5			5	253	175	-206	271	14		788	-432	-583	725	14		788	-432	-583	725		
			1	399	-378	-82	387			6	1061	968	431	1060	15		504	-220	413	468	15		504	-220	413	468		
			2	946	646	-636	907			7	535	-364	397	539	16		346	-257	229	344	16		346	-257	229	344		
			3	640	38	-641	643			8	420	332	-285	457	17		284	-44	272	276	17		284	-44	272	276		
			4	936	881	567	1048			9	353	194	-69	206	18		189	19	232	232	18		189	19	232	232		
			5	273	-158	-115	195			10	1051	-1067	68	1070	8	3	0	651	-629	0	629	8	3	0	651	-629	0	629
			6	746	335	-554	648			11	910	-1	373	133	1		1524	-1466	-650	1604	1		1524	-1466	-650	1604		
			7	516	-155	-433	460			12	1218	1254	376	1310	2		883	-514	831	977	2		883	-514	831	977		
			8	1261	1191	221	1212			13	263	214	173	275	3		1146	732	-903	1175	3		1146	732	-903	1175		
			9	862	222	-718	751			14	621	654	-133	679	4		883	-804	399	897	4		883	-804	399	897		
			10	525	431	-54	434			15	200	-85	-203	220	5		977	171	-928	944	5		977	171	-928	944		
			11	604	169	-468	497			16	325	355	100	369	6		1356	-1284	-642	1435	6		1356	-1284	-642	1435		
			12	673	630	-51	632			17	<220	16	14	21	7		704	-13	-768	768	7		704	-13	-768	768		
			13	284	234	174	292			18	442	380	45	382	8		410	-414	10	414	8		410	-414	10	414		
13	2	0	284	-284	0	289	19			19	148	-99	140	172	9		200	-225	134	262	9		200	-225	134	262		
			1	516	-408	-330	525			20	473	262	366	450	10		851	-785	-5	785	10		851	-785	-5	785		
			2	346	55	-319	324			21	210	241	44	245	11		358	-129	-320	345	11		358	-129	-320	345		
			3	<241	76	60	97			4	3	0	525	-477	0		442	-375	89	386	0		442	-375	89	386		
			4	410	300	-38	374			1	2439	-111	2438	2440	13		883	-98	-83	858	13		883	-98	-83	858		
			5	<232	24	-7	25			2	241	154	18	155	14		831	-831	159	846	14		831	-831	159	846		
			6	494	450	-36	452			3	925	-143	854	865	15		778	-324	-666	741	15		778	-324	-666	741		
			7	568	66	-543	547			4	1061	-1002	-225	1027	16		516	-220	-423	477	16		516	-220	-423	477		
			8	346	-258	193	322			5	1619	78	1627	1629	17		210	-55	-240	246	17		210	-55	-240	246		
			9	315	271	153	311			6	1187	-665	979	1183	9	3	0	420	470	0	470	9	3	0	420	470	0	470
			10	547	481	-47	483			7	1020	-346	924	987	1		662	301	-596	667	1		662	301	-596	667		
			11	<136	-133	-71	151			8	936	-897	338	959	2		726	-740	381	832	2		726	-740	381	832		
14	2	0	<189	18	0	18	9			9	1703	-1431	738	1610	3		389	160	-383	415	3		389	160	-383	415		
			1	430	-266	331	424			10	726	-455	-383	579	4		915	-825	-401	917	4		915	-825	-401	917		
			2	862	804	-45	805			11	2375	2432	1159	2694	5		693	184	670	694	5		693	184	670	694		
			3	683	236	121	266			12	757	-595	-420	728	6		346	-125	-250	279	6		346	-125	-250	279		
			4	325	319	138	348			13	1061	-318	1001	1050	7		525	533	-41	535	7		525	533	-41	535		
			6	599	621	135	636			14	200	-106	-183	211	8		389	-86	413	422	8		389	-86	413	422		
			7	483	415	273	497			15	893	-105	942	947	9		442	-454	154	479	9		442	-454	154	479		
			8	263	238	-155	284			16	452	-90	404	414	10		714	-516	-393	649	10		714	-516	-393	649		
			2	2448	2312	-1045	2538			17	925	-34	790	791	11		<210	-114	71	134	11		<210	-114	71	134		
			4	967	878	-89	882			18	<189	-70	171	184	12		430	-34	-341	343	12		430	-34	-341	343		
			6	1902	1970	-190	1979			19	893	-42	926	927	13		810	-626	-302	695	13		810	-626	-302	695		
			8	1629	1650	164	1658			20	378	-351	116	369	14		557	-393	266	475	14		557	-393	266	475		
			10	2407	2627	-204	2635			5	3	0	473	0	473	15		<136	40	104	112	15		<136	40	104	112	
			12	337	321	8	321			1	841	317	712</															

TABLE 8. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	100 <i>kF</i> _o	100 <i>A</i> _c	100 <i>B</i> _c	100 <i>F</i> _c	
12	3	2	389	-314	-157	351	12	3	8	389	193	289	348	13	3	3	136	121	-182	219	
		3	640	-55	604	607			9	158	28	-197	199			4	378	376	54	380	
		4	1777	1577	-564	1675			10	430	483	-67	488			5	315	219	-247	350	
		5	831	-184	-711	734	13	3	0	305	326	0	326			6	179	140	-151	206	
		6	494	537	69	541			1	168	-42	-192	196								
		7	420	-19	-372	372			2	<136	-134	26	136								

O, carbons 2, 1, 7, and N (ring C). It is reasonable to interpret this peak as the hydroxylic hydrogen since the O-H length and the COH angle compare well with the corresponding values in salicylic acid⁹ (1.04 Å, 109°).

The electron-density distribution in the room-temperature structure is more diffuse; $\delta(xz)$ shows an elongated peak of maximum height 0.25 e/Å² which in the three-dimensional plane sections (Fig. 4) is resolved into two peak systems. The first, M_1 (>0.25 e/Å³) lies at 0.89 Å from the oxygen, and in the plane of ring C such that the angle C₂OM₁ is 106.5°. The second peak, M_2 (<0.25 e/Å³) lies at 1.65 Å from the oxygen, at an angle C₂OM₂ of 85°, and approximately 1 Å above ring C. M_1 may be assigned to the hydroxylic hydrogen; the meaning to be attached to M_2 is highly problematical.

One may not unjustifiably query the significance of these electron-density determinations on the grounds of inadequate experimental measurements (*e.g.*, photographic as well as only partial *hkl* data) and possibly incomplete refinement (diagonal approximation of the least-squares procedure used here). However, we have chosen to submit these findings on the grounds that they represent our experimental results, that the observed "hydrogen" positions are not unreasonable as regards bond lengths and angles, that they are partly confirmed by similar results of our analysis of 2-chloro-*N*-salicylideneaniline (Part X, following Paper) and by our partial refinement¹⁰ of the structure of 5-chlorosalicylaldehyde.¹¹

The positional and thermal parameters derived from the two sets of data, and their standard deviations estimated according to Part V⁵ are listed in Tables 2–5. Bond lengths and bond angles are given in Figs. 5a and b; the C-H bond lengths and angles are found in Table 6. The best planes of the three rings C₁–C₆ (A), C₈–C₁₃ (B), and OC₂C₁C₇N (C) were computed according to Schomaker *et al.*¹² Table 7 lists their equations, as well as the distances of the substituent and ring atoms from these planes. Table 8 contains *hkl*, 100 *kF*_o, 100 *A*_c, 100 *B*_c, and 100 *F*_c. Reflections to which zero weight was attached are asterisked; all unobserved reflections are marked <*kF*_o.

Analysis of Thermal Vibrations.—The anisotropic thermal parameters obtained from the least-squares refinement were analysed for the translational and librational components of rigid-body motion by Cruickshank's method,¹³ modified by Hirshfeld¹⁴ for molecules in which the point of intersection of the axes of libration is not known. The magnitudes and direction cosines, with respect to the unit-cell axes, of the principal axes of the ellipsoids of motion are given in Table 9 which also lists the co-ordinates of the centre of libration. The ellipsoids of libration of the low-temperature and room-temperature structures agree closely both in the orientation and relative magnitudes of their respective principal axes. (The weakest principal component of libration of the low-temperature structure is negative; however, its magnitude is negligible.) The dominant mode of libration occurs about an axis lying along the length of the molecule; the molecule oscillates about an axis nearly normal to its plane. Its translational motion is almost isotropic.

The r.m.s. discrepancy of the rigid-body analysis equals $[\sum \Delta u_{ij}^2 / (n - s)]^{1/2}$, where *n* is the total number of thermal parameters and *s* the number of independent parameters,

⁹ Cochran, *Acta Cryst.*, 1953, **6**, 260.

¹⁰ Leiserowitz and Schmidt, to be published.

¹¹ Simonsen, Pflugger, and Thompson, *Acta Cryst.*, 1961, **14**, 269.

¹² Schomaker, Waser, Marsh, and Bergman, *Acta Cryst.*, 1959, **12**, 600.

¹³ Cruickshank, *Acta Cryst.*, 1956, **9**, 754, 757.

¹⁴ Hirshfeld, Sandler, and Schmidt, *J.*, 1963, 2108.

TABLE 9.

Principal components of rigid-body vibrations and their direction cosines relative to the crystal axes. Co-ordinates of centres of libration.

	Low temperature				Room temperature		
	<i>a</i>	<i>b</i>	<i>c</i>		<i>a</i>	<i>b</i>	
$T_1^2 = 0.0094 \text{ \AA}^2$...	-0.9845	0.1601	-0.0719	$T_1^2 = 0.0266 \text{ \AA}^2$...	-0.9555	-0.1955	0.2211
$T_2^2 = 0.0078$	0.1749	0.9286	0.3273	$T_2^2 = 0.0440$	-0.0146	-0.7168	-0.6971
$T_3^2 = 0.0125$	0.0143	-0.3348	-0.9422	$T_3^2 = 0.0326$	0.2948	-0.6693	0.6820
$\omega_1^2 = -0.00008 \text{ rad.}^2$	-0.9908	-0.0147	-0.1345	$\omega_1^2 = 0.00028 \text{ rad.}^2$	-0.9808	0.0352	-0.1916
$\omega_2^2 = 0.00028$	0.0599	0.8434	-0.5340	$\omega_2^2 = 0.00131$	0.1478	0.7752	-0.6143
$\omega_3^2 = 0.00139$	0.1213	-0.5371	-0.8347	$\omega_3^2 = 0.00738$	0.1269	-0.6308	-0.7655
Libration centre at	<i>x</i>	<i>y</i>	<i>z</i>				
Low temperature...	2.671	2.624	3.800				
Room temperature	2.237	2.758	3.719				

i.e., six translational, six librational, and three defining the origin of libration. The r.m.s. discrepancy equals 0.0028 and 0.0042 \AA^2 for the low-temperature and room-temperature structures, respectively. The corresponding r.m.s. standard deviations of thermal vibration from the least-squares procedure are 0.0021 and 0.0028 \AA^2 . Because the least-squares estimate is likely to be too small the differences between the two sets of r.m.s. values cannot be regarded as evidence for internal vibrations.

Table 10 shows that the agreement between the observed and calculated thermal

TABLE 10.

Thermal parameters (\AA^2) derived from the analysis of rigid-body motion.

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{23}	U_{13}
Low temperature						
C_1	0.0096	0.0085	0.0120	-0.0004	0.0014	0.0000
C_2	0.0100	0.0092	0.0124	0.0003	0.0012	0.0001
C_3	0.0116	0.0099	0.0127	0.0012	0.0011	0.0005
C_4	0.0133	0.0079	0.0117	0.0001	0.0020	0.0007
C_5	0.0125	0.0090	0.0122	-0.0012	0.0012	0.0000
C_6	0.0105	0.0103	0.0129	-0.0008	0.0005	-0.0004
C_7	0.0095	0.0091	0.0124	0.0000	0.0012	0.0000
C_8	0.0119	0.0082	0.0118	0.0000	0.0018	0.0005
C_9	0.0130	0.0109	0.0132	0.0022	0.0009	0.0006
C_{10}	0.0163	0.0114	0.0134	0.0035	0.0011	0.0013
C_{11}	0.0191	0.0079	0.0115	0.0015	0.0026	0.0018
C_{12}	0.0174	0.0075	0.0113	-0.0011	0.0023	0.0010
C_{13}	0.0137	0.0083	0.0118	-0.0012	0.0016	0.0003
N	0.0100	0.0083	0.0119	-0.0003	0.0015	0.0001
O	0.0096	0.0123	0.0140	0.0007	-0.0003	-0.0005
Cl	0.0156	0.0127	0.0141	-0.0026	-0.0007	-0.0011
Room temperature						
C_1	0.0281	0.0380	0.0377	-0.0010	0.0056	0.0016
C_2	0.0295	0.0476	0.0451	0.0030	0.0003	0.0009
C_3	0.0361	0.0531	0.0503	0.0069	-0.0028	0.0032
C_4	0.0430	0.0424	0.0424	0.0022	0.0028	0.0057
C_5	0.0405	0.0415	0.0399	-0.0029	0.0028	0.0018
C_6	0.0321	0.0431	0.0408	-0.0023	0.0022	0.0001
C_7	0.0275	0.0385	0.0381	-0.0008	0.0053	0.0015
C_8	0.0376	0.0394	0.0389	-0.0017	0.0043	0.0027
C_9	0.0410	0.0456	0.0450	0.0044	0.0012	0.0052
C_{10}	0.0544	0.0499	0.0496	0.0076	-0.0015	0.0090
C_{11}	0.0666	0.0430	0.0436	0.0006	0.0013	0.0095
C_{12}	0.0619	0.0470	0.0436	-0.0059	-0.0016	0.0026
C_{13}	0.0464	0.0481	0.0438	-0.0049	-0.0016	-0.0004
N	0.0301	0.0397	0.0387	-0.0016	0.0044	0.0010
O	0.0290	0.0659	0.0575	0.0059	-0.0102	-0.0039
Cl	0.0547	0.0586	0.0500	-0.0064	-0.0082	-0.0040

vibrations is not unreasonable. However, two points must be stressed: since the molecule is nearly planar and the nitrogen atom likely to be strongly conjugated to its phenyl group we may assume *a priori* that the two benzene rings librate in phase rather than independently. However, while the major librating axis has been found to lie along the length of the molecule a decision between in-phase and independent vibrations cannot be based on the rigid-body analysis just presented. Secondly, since the anomalous-dispersion correction of the chlorine scattering function was not applied in the least-squares analysis the temperature factor of chlorine must be regarded with reserve.

Small oscillations of r.m.s. amplitude θ about an axis cause the calculated position of an atom, whose distance from this axis is r , to be displaced towards the libration axis by approximately $\frac{1}{2}r\theta^2$. The maximum changes in bond lengths have been found to be 0.001 and 0.006 Å for the low- and room-temperature data, respectively. The libration-corrected bond lengths of the room-temperature structure are listed in Table II.

TABLE II.

Bond lengths (room temperature) corrected for thermal vibration.		Standard deviation of bond lengths (Å).	
Bond	Length (Å)	Low temp.	Room temp.
C ₁ -C ₂	1.423	0.0069	0.0059
C ₂ -C ₃	1.399	0.0071	0.0062
C ₃ -C ₄	1.366	0.0071	0.0064
C ₄ -C ₅	1.411	0.0074	0.0063
C ₅ -C ₆	1.361	0.0073	0.0059
C ₆ -C ₁	1.411	0.0069	0.0057
C ₁ -C ₇	1.445	0.0070	0.0058
C ₇ -N	1.273	0.0066	0.0053
N-C ₈	1.420	0.0068	0.0053
C ₈ -C ₉	1.415	0.0076	0.0062
C ₉ -C ₁₀	1.368	0.0073	0.0069
C ₁₀ -C ₁₁	1.390	0.0074	0.0072
C ₁₁ -C ₁₂	1.377	0.0076	0.0074
C ₁₂ -C ₁₃	1.405	0.0074	0.0071
C ₈ -C ₁₃	1.391	0.0075	0.0063
C ₅ -Cl	1.759	0.0056	0.0046
C ₂ -O	1.356	0.0060	0.0054

The results of this analysis (molecular shape, packing arrangement, bond lengths, and angles) are discussed in Part X (following paper) where they are compared with the data on 2-chloro-*N*-salicylideneaniline.

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